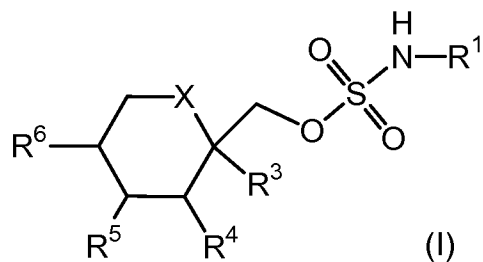


In the Claims:

This listing of claims will replace all prior versions, and listings, of claims in the Application.

1. (Currently Amended) A continuous process for the preparation of a compound of formula (I)

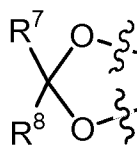


wherein

X is selected from CH₂ or O;

R¹ is selected from the group consisting of hydrogen and C₁₋₄alkyl;

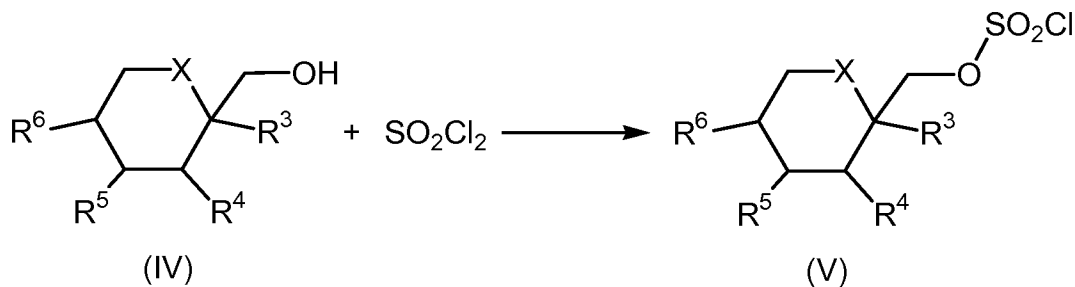
R³, R⁴, R⁵ and R⁶ are each independently selected from hydrogen or lower alkyl and, when X is CH₂, R⁵ and R⁶ may be alkene groups joined to form a benzene ring and, when X is O, R³ and R⁴ and/or R⁵ and R⁶ together may be a methylenedioxy group of the formula:



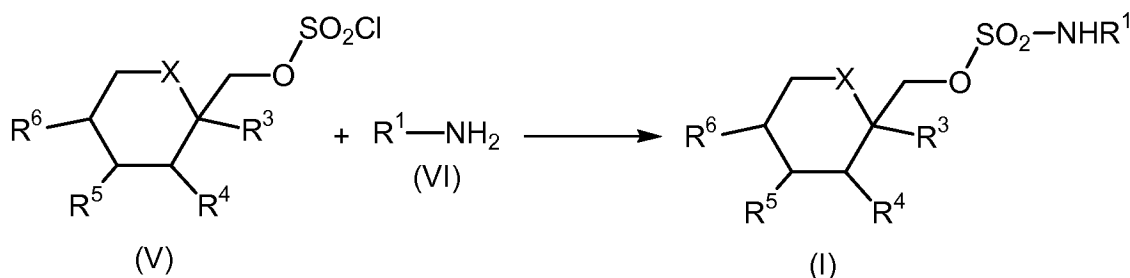
wherein

R⁷ and R⁸ are same or different and are hydrogen, lower alkyl or are alkyl and are joined to form a cyclopentyl or cyclohexyl ring;

comprising



(A) continuously reacting a suitably substituted compound of formula (IV) with sulfonyl chloride;
 in the presence of an organic or inorganic base;
 in a first organic solvent ~~selected from a cyclic ether, or a straight or branched chain dialkyl ether;~~
~~wherein the first organic solvent is other than t-butyl methyl ether or tetrahydrofuran;~~
 to form the corresponding compound of formula (V); and



(B) continuously reacting the compound of formula (V) with a suitably substituted compound of formula (VI);
 in a second organic solvent ~~which is the same as the first organic solvent used in step (A);~~
wherein the first organic solvent and the second organic solvent are each glyme;
 to form the corresponding compound of formula (I).

2. (Original) The process as in Claim 1, wherein the organic or inorganic base is an organic base.

3. (Original) The process as in Claim 2, wherein the organic base is pyridine.

Claims 4-5. (Canceled)

6. (Canceled)

7. (Original) The process as in Claim 1, wherein the sulfonyl chloride is present in an amount greater than about 0.9 moles per mole of the compound of formula (IV).

8. (Original) The process as in Claim 1, wherein the base is present in an amount greater than about 1 molar equivalent of the compound of formula (IV).

9. (Original) The process as in Claim 8, wherein the molar ratio of the compound of formula (IV) to the base is at least about 1:1.05.

10. (Original) The process as in Claim 1, wherein the temperature of the reaction in Step (A) is less than about 50°C.

Claim 11. (Canceled)

12. (Original) The process as in Claim 1, wherein the compound of formula (VI) is present in an amount greater than about 1 molar equivalent of the compound of formula (V).

13. (Original) The process as in Claim 12, wherein the molar ratio of the compound of formula (VI) to the compound of formula (V) is at least about 2:1.

14. (Original) The process as in Claim 1, wherein the compound of formula (VI) is ammonia and the ammonia is fed into the reactor at a pressure of about 19 psia.

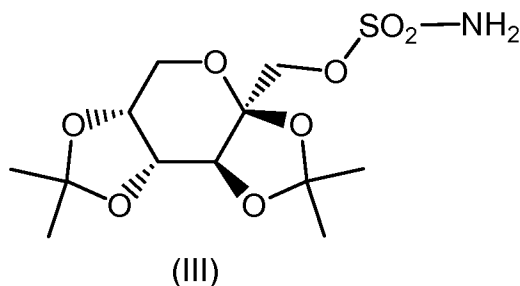
15. (Original) The process as in Claim 1, wherein the temperature of the reaction in Step (B) is in the range of from about -30 to about 50°C.

16. (Original) The process as in Claim 1, wherein the compound of formula (V) is formed in a solution comprising the compound of formula (V) and the first organic solvent.

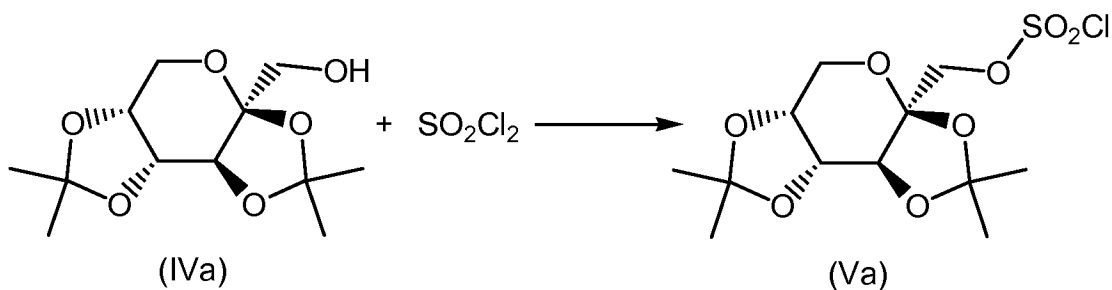
17. (Original) The process as in Claim 16, wherein the sulfuryl chloride is reacted in amount equal to about 1 equivalent relative to the compound of formula (IV), further comprising concentrating the solution comprising the compound of formula (V) and the first organic solvent to remove at least about 20% of the solvent mass.
18. (Original) The process as in Claim 16, wherein the sulfuryl chloride is reacted in amount greater than about 1 equivalent relative to the compound of formula (IV), further comprising concentrating the solution comprising the compound of formula (V) and the first organic solvent to remove at least 70% of the solvent mass.
19. (Original) The process as in Claim 16, further comprising concentrating the solution comprising the compound of formula (V) and the first organic solvent to an oil.
20. (Original) The process as in Claim 16, further comprising treating the solution comprising the compound of formula (V) and the first organic solvent to remove volatiles.
21. (Original) The process as in Claim 1, wherein step (A) and step (B) are each run in a continuous stirred tank reactor.

Claims 22-25. (Canceled)

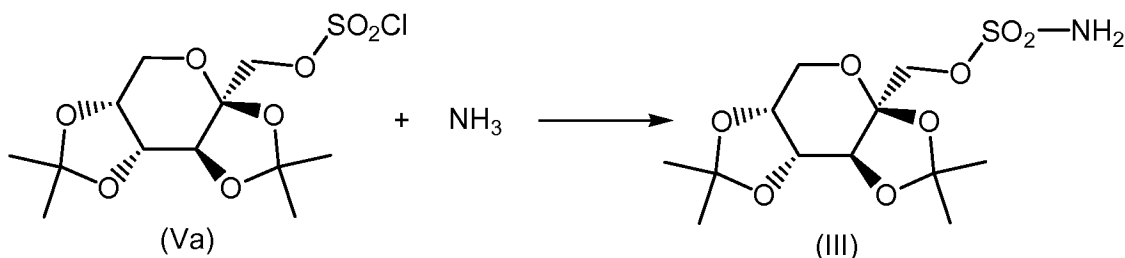
26. (Currently Amended) A continuous process for the preparation of a compound of formula (III)



comprising



(Aa) continuously reacting a compound of formula (IVa) with sulfuryl chloride;
 in the presence of an organic or inorganic base;
 in a first organic solvent ~~selected from a cyclic ether, or a straight or branched chain dialkyl ether;~~
~~wherein the first organic solvent is other than t butyl methyl ether or tetrahydrofuran;~~
 to form the corresponding compound of formula (Va); and



(Ba) continuously reacting the compound of formula (Va) with ammonia;
 in a second organic solvent ~~which is the same as the first organic solvent used in step (Aa);~~
wherein the first organic solvent and the second organic solvent are each glyme;
 to form the corresponding compound of formula (III).

27. (Original) The process as in Claim 26, wherein the organic or inorganic base is an organic base.

28. (Original) The process as in Claim 27, wherein the organic base is pyridine.

Claims 29-31. (Canceled)

32. (Original) The process as in Claim 26 wherein the sulfonyl chloride is present in an amount greater than about 0.9 moles per mole of the compound of formula (IVa).
33. (Original) The process as in Claim 26, wherein the base is present in an amount greater than about 1 molar equivalent of the compound of formula (IVa).
34. (Original) The process as in Claim 33, wherein the molar ratio of the compound of formula (IVa) to the base is at least about 1:1.05.
35. (Original) The process as in Claim 26, wherein the temperature of the reaction in Step (Aa) is less than about 50°C.
- Claim 36. (Canceled)
37. (Original) The process as in Claim 26, wherein the ammonia is present in an amount greater than about 1 molar equivalent of the compound of formula (Va).
38. (Original) The process as in Claim 37, wherein the molar ratio of the ammonia to the compound of formula (Va) is at least about 2:1.
39. (Original) The process as in Claim 26, wherein ammonia is fed into the reactor at a pressure of about 19 psia.
40. (Original) The process as in Claim 26, wherein the temperature of the reaction in Step (Ba) is in the range of from about -30 to about 50°C.
41. (Original) The process as in Claim 26, wherein the compound of formula (Va) is formed in a solution comprising the compound of formula (Va) and the first organic solvent.

42. (Original) The process as in Claim 41, wherein the sulfuryl chloride is reacted in amount equal to about 1 equivalent relative to the compound of formula (IVa), further comprising concentrating the solution comprising the compound of formula (Va) and the first organic solvent to remove at least about 20% of the solvent mass.
43. (Original) The process as in Claim 41, wherein the sulfuryl chloride is reacted in amount greater than about 1 equivalent relative to the compound of formula (IVa), further comprising concentrating the solution comprising the compound of formula (Va) and the first organic solvent to remove at least 70% of the solvent mass.
44. (Original) The process as in Claim 41, further comprising concentrating the solution comprising the compound of formula (Va) and the first organic solvent to an oil.
45. (Original) The process as in Claim 41, further comprising treating the solution comprising the compound of formula (Va) and the first organic solvent to remove volatiles.
46. (Original) The process as in Claim 26, wherein step (Bb) is run in a continuous stirred tank reactor.
47. (Original) The process as in Claim 26, wherein step (Aa) and step (Ba) are each run in a continuous stirred tank reactor.